Mar-Apr 1997

Bicyclic [b]-Heteroannulated Pyridazine Derivatives. 4 [1]. Cyclization Reactions of 4-Aryltetrahydropyridazine-3,6-dione 3-Hydrazones with Some Keto Esters

Jerzy Lange*[a], Janina Karolak-Wojciechowska [b], Elzbieta Pytlewska [a], Jan Plenkiewicz [a], Tomasz Kulinski [a], and Slawomir Rump [c]

[a] Department of Chemistry, University of Technology, Noakowskiego 3, PL-00664 Warsaw, Poland
[b] Institute of General and Ecological Chemistry, Technical University, Zwirki 36, PL-90924 Lodz, Poland
[c] Military Institute of Hygiene and Epidemiology, Kozielska 4, PL-01163 Warsaw, Poland
Received January 8, 1996
Revised October 15, 1996

Ethoxycarbonylalkylidene derivatives 2 and 6 of the title hydrazones were obtained in the reaction with ethyl pyruvate or ethyl aroylformate and ethyl acetoacetate, respectively, in methanol. Both compounds were mixtures of geometric isomers with high predominance of one of them. Nmr spectroscopy revealed an unexpected magnetic non-equivalence of the CH₂ protons in the ester ethyl group of the major isomer of 6. On heating (-200°) in an inert medium or on refluxing in ethanolic sodium ethoxide 2 cyclized to the corresponding pyridazine[6,1-c]-triazines 4, whereas 6 formed pyrazolylpyridazines 7. The structure of the latter was unambigously established by X-ray analysis. Alkylation of 4a with benzyl bromide in the presence of tetrabutylammonium bromide occurred selectively on the pyridazine N atom.

J. Heterocyclic Chem., 34, 389 (1997).

Triazolo[4,3-b]pyridazines are well known as ligands of the benzodiazepine receptor [2-5]. The most active compound of this type, 3-methyl-6-(3-trifluoromethylphenyl)triazolo[4,3-b]pyridazine, often referred to as CL 218,872, was found *in vitro* to be an effective inhibitor of 3 H-diazepam and 3 H-flunitrazepam binding to the receptor [2,4,6-8]. Although, despite some early expectations, not introduced as a drug, it is still the object of pharmacological interest as a reference ligand with highly selective affinity for the ω_1 subtype of the benzodiazepine receptor [9,10].

Considering the convenient availability of the 4-aryl-substituted tetrahydropyridazine-3,6-dione 3-hydrazones [11-13], we report now on the use of these versatile reagents for the preparation of some bicyclic compounds with potential biological activity in the reaction with keto esters.

4-Aryltetrahydropyridazine-3,6-dione 3-hydrazones 1a-d, obtained in the reaction of ethyl 3-aryl-3-cyanopropionate with hydrazine hydrate [11,12], reacted at room temperature with ethyl pyruvate and ethyl aroylformates to give ethyl 2-[(4-aryl-6-oxo-3-tetrahydropyridazinylidene)hydrazono]propanoates 2a-d and ethyl α -[(4-phenyl-6-oxo-3-tetrahydropyridazinylidene)hydrazono]-arylacetates 2e-f, respectively, in the yield of 40-50% (Scheme 1) [13]. In methanol at reflux temperature the conversion was higher (>70%) but purification was required to remove the cyclocondensation by-products.

In the ¹H-nmr spectra of **2a-d** some signals of the sidechain protons were split thus indicating presence of two geometric isomers involving the C=N bond. This was seen most clearly with the singlet signal of the CH₃ group (2.17 in **2a**) which was accompanied with a low intensity (approximately 15% of the intensity of the former) singlet at δ 2.09. No attempts were made to separate the isomers. Cyclization of 2 to the pyridazino[6,1-c][1,2,4]triazine derivatives 4a-f was effected most conveniently by heating at approximately 200°, preferably in a high boiling solvent such as 2-(2-ethoxyethoxy)ethanol or ethylene glycol. A similar cyclization of the acids 3a-d, prepared from 2a-d by routine sodium ethoxide alcoholysis, gave much lower yields since several recrystallizations were required to remove by-products. The bicyclic compounds 4e-f were also obtained, although in moderate yields, in a direct condensation of 1a with the appropriate ethyl aroylformate in refluxing 2-(2-ethoxyethoxy)ethanol or ethylene glycol.

Alkylation of **4a** with benzyl bromide in DMF in the presence of tetrabutylammonium bromide and potassium hydroxide yielded a compound for which the structure of the 6-benzyl derivative **5** was assigned. The assignment was based on the ¹H-nmr spectrum in which the benzyl CH₂ protons appeared as a double doublet (AB splitting pattern) owing to chirality of the tetrahedral N6 atom. The formation of the *O*-benzylated compound was not observed [14].

The reaction of 1a-d with ethyl acetoacetate yielded the corresponding ethyl 3-[(4-aryl-6-oxotetrahydro-3-pyridazinylidene)hydrazono]butanoates 6a-d. In this case, too, two sets of signals (approximately 6:1 intensity ratio) observed in the $^1\mathrm{H}$ -nmr spectra of 6 indicated presence of geometric isomers involving the C=N bond formed in the condensation reaction. Moreover, an unexpected and interesting magnetic non-equivalence of the CH2 protons in the ester ethyl group (δ 4.08 and 4.10 ppm, a quartet of double-doublet signals of unequal intensities resulting from the combination of the AB- and AX3-pattern spin couplings) was observed in the spectrum of the major isomer of 6a. Both protons have practically identical coupling constants

J. Lange, J. Karolak-Wojciechowska, E. Pytlewska, J. Plenkiewicz, T. Kulinski, and S. Rump

with the methyl group (J = 7.2 Hz) and are coupled with one another with J = 11.0 Hz. Since there are distinct proton-donating and -accepting sites in the molecule, intraand/or intermolecular hydrogen bonds may be responsible for this non-equivalence. For instance, if one considers the (Z)-configuration on the C=N bond formed in the condensation reaction, rotation of the ethoxycarbonylmethyl fragment about the C-C bond can bring the ester alkoxy and carbonyl oxygen atoms quite close (approximately 2.5 Å) to the hydrogen atom attached to N1. The C=N bond linking the chain with the pyridazine core may be involved in a prototropic rearrangement (N2 = C3 or C3 = exocyclic N) [12] so consideration of the (Z)-(E) isomerism on this bond is hardly possible at present. The assignments of the ¹H- and ¹³C-nmr signals in the spectrum of both isomers of 6a are given in detail in the Experimental. The spectra of 6b and 6d show a similar non-equivalence but the ester group signals of the major and minor isomers closely overlap each other and therefore they are reported as multiplets. No such non-equivalence was noted in the case of the methoxy derivative 6c.

When heated with ethanolic sodium ethoxide 6 cyclized with elimination of ethanol. Subsequent acidification of the initially obtained sodium derivatives yielded the corresponding *aci*-forms. Considering the possible tautomerism of 6 [12], the cyclization could have occurred either on N2 with the formation of a 7-membered triazepine ring fused with the pyridazine or on the hydrazine nitrogen atom with

the formation of a 5-membered pyrazole ring not fused with the pyridazine. Earlier reports on similar cyclocondensations with various nitrogen heterocycles are rather confusing in that point. Thus, the reaction of ethyl acetoacetate with a derivative of 2-hydrazinothiazole was reported to yield the corresponding thiazolotriazepine [15]. On the other hand, there are several reports on the formation of pyrazolyl heterocycles in related reactions [16-19]. The latter possibility was certainly much more likely but it was not possible to get a definite answer by examination of the ir and nmr spectra of 7. Since the non-aromatic character of the pyridazine ring distinguished 6 from the objects of those earlier investigations, one of the cyclic compounds, 7d obtained from 6d, was subject to an X-ray analysis which confirmed the formation of a pyrazolone. The ORTEP drawing of this structure with atom numbering and selected intermolecular hydrogen bonds is shown as Figure 1 while the non-hydrogen fractional atomic coordinates are collected in Table 1, and bond length and angle data, in Table 2 [20].

In the folded pyridazine ring of 7d, the sp^3 -hybridized C5 atom is located 0.567(6)Å above the plane defined by the two nitrogen atoms and the two sp^2 -hybridized carbon atoms C3 and C6 (plane i). The dihedral angle between this plane and the phenyl ring was found to be $86.9(2)^\circ$. The pyrazolone ring is roughly planar and is twisted by $77.1(2)^\circ$ with respect to the i plane. Standard bond lengths and valence angles characterize the pyridazine and phenyl rings.

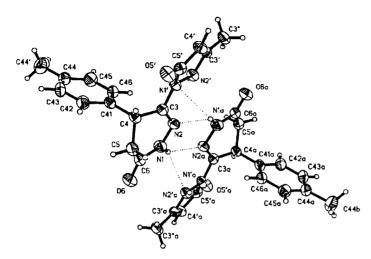


Figure 1. Molecular structure of the 7d crystal presented as a centrosymmetric hydrogen-bonded dimer. For the sake of drawing clarity, the strong hydrogen bond (N2'-H2'...O5') extending along the x axis is not shown.

It was possible to locate the proton bonded to N2' in the pyrazolone moiety using X-ray techniques. The C3'-C4' distance (1.356(8) Å) indicates the presence of a double bond between these atoms. Other endocyclic bonds are relatively short, whereas the exocyclic C5'-O5' bond is slightly longer than expected (1.260(7) Å). The hybridization of the N2' atom is close to sp^2 (the sum of valence angles around this atom is 353.9). Crystals of 7d show therefore a possible tautomerism of this compound (Scheme 2).

Table 1

Fractional Atomic Coordinates (x 10^{-4}) and Equivalent Isotropic Displacement Parameters (\mathring{A}^2 x 10^3) for the Non-hydrogen Atoms of 7d. U_{ea} defined as 1/3 of the trace of the orthogonalized U_{ii} tensor

Atom	x/a	y/b	z/c	\mathbf{U}_{eq}
C3	4144(10)	-1266(6)	-854(5)	30(1)
N2	2357(9)	-585(5)	-524(4)	36(1)
N1	1502(9)	565(5)	-1241(4)	40(1)
C6	2759(11)	1215(6)	-2120(5)	37(1)
O6	1857(8)	2221(5)	-2694(4)	58(1)
C5	5200(10)	660(6)	-2256(5)	38(2)
C4	5501(10)	-878(6)	-1931(4)	32(1)
N1'	4729(8)	-2520(5)	-182(4)	34(1)
N2'	2976(8)	-3038(5)	573(4)	34(1)
C3'	4044(11)	-4115(5)	1248(4)	33(1)
C3"	2557(11)	-4940(6)	2080(5)	43(2)
C4'	6382(10)	-4203(6)	1015(5)	38(2)
C5'	6879(10)	-3177(6)	113(5)	33(1)
O5'	8774(7)	-2829(4)	-371(3)	44(1)
C41	4758(10)	-1600(6)	-2819(4)	33(1)
C42	6104(11)	-1581(6)	-3790(5)	44(2)
C43	5520(12)	-2222(6)	-4628(5)	48(2)
C44	3480(12)	-2855(6)	-4518(5)	44(1)
C45	2121(12)	-2863(7)	-3553(5)	48(2)
C46	2682(11)	-2208(6)	-2722(5)	38(1)
C44'	2825(14)	-3550(8)	-5422(6)	71(2)

Table 2
Bond Lengths (Å) and Bond Angles (deg) in 7d

		5 \ 2,	
C(3)-N(2)	1.266(8)	N(2)-C(3)-C(4)	124.5(5)
C(3)-C(4)	1.503(8)	N(2)-C(3)-N(1')	115.7(5)
C(3)-N(1')	1.412(7)	C(4)-C(3)-N(1')	119.6(5)
N(2)-N(1)	1.400(6)	C(3)-N(2)-N(1)	117.7(5)
N(1)-C(6)	1.365(8)	N(2)-N(1)-C(6)	124.5(5)
C(6)-O(6)	1.224(7)	N(1)-C(6)-O(6)	119.5(6)
C(6)-C(5)	1.472(8)	N(1)-C(6)-C(5)	115.3(5)
C(5)-C(4)	1.536(8)	O(6)-C(6)-C(5)	125.1(6)
C(4)-C(41)	1.531(9)	C(6)-C(5)-C(4)	112.6(5)
N(1')-N(2')	1.402(6)	C(3)-C(4)-C(5)	108.2(5)
N(1')-C(5')	1.387(7)	C(3)-C(4)-C(41)	111.2(5)
N(2')-C(3')	1.369(7)	C(5)-C(4)-C(41)	113.0(5)
C(3')-C(3")	1.484(8)	C(3)-N(1')-N(2')	116.5(5)
C(3')-C(4')	1.356(8)	C(3)-N(1')-C(5')	131.0(5)
C(4')-C(5')	1.415(8)	N(2')-N(1')-C(5')	109.6(5)
C(5')-O(5')	1.260(7)	N(1')-N(2')-C(3')	105.9(4)
C(41)-C(42)	1.379(8)	N(2')-C(3')-C(3")	117.9(5)
C(41)-C(46)	1.391(9)	N(2')-C(3')-C(4')	110.3(5)
C(42)-C(43)	1.394(10)	C(3")-C(3')-C(4')	131.8(5)
C(43)-C(44)	1.385(10)	C(3')-C(4')-C(5')	108.4(5)
C(44)-C(45)	1.377(9)	N(1')-C(5')-C(4')	105.4(5)
C(44)-C(44')	1.511(11)	N(1')-C(5')-O(5')	122.7(5)
C(45)-C(46)	1.392(10)	C(4')-C(5')-O(5')	131.9(6)
		C(4)-C(41)-C(42)	119.4(5)
		C(4)-C(41)-C(46)	122.6(5)
		C(42)-C(41)-C(46)	117.8(5)
		C(41)-C(42)-C(43)	122.1(6)
		C(42)-C(43)-C(44)	119.9(6)
		C(43)-C(44)-C(45)	118.0(6)
		C(43)-C(44)-C(44')	120.4(6)
		C(45)-C(44)-C(44')	121.5(6)
		C(44)-C(45)-C(46)	122.1(6)
		C(41)-C(46)-C(45)	119.8(6)

From the thermodynamic point of view the form C, shown in Scheme 2, appears to be the most stable. The heat of its formation, calculated by the PM3-MNDO method, is 8.09 kcal/mole as compared with 12.13 and 18.84 kcal/mole for tautomers A and B, respectively [21]. In the crystalline state, however, presumably because of the presence of the

NH group which favors the formation of intermolecular hydrogen bonds, tautomer A predominates. Two bifurcate hydrogen bonds (N1-H1....N2 and N1-H1....N2') join two molecules of 7d together which forms a dimer with the molecules related by a center of symmetry (Figure 1). Another strong hydrogen bond (N2'-H2'....O5'), which for the sake of clarity is not shown in Figure 1, links one dimer to another dimer separated by a cell translation along the x axis (Table 3). It is worth noting that the latter hydrogen bond may be regarded as a kind of a bridge through which the H2' proton can pass from N2' to O5' and vice versa in the interconversion of the A and B tautomers.

Table 3
Hydrogen Bonding Contacts (Å) and Angles (deg)

Х-НҮ	х-н	НУ	XY	Х-НҮ	Symmetry translation of Y
N1-H1N2	0.95(7)	2.217(6)	3.002(7)	139.7(4)	-x, -y, <i>-</i> z
N1-H1N2'	0.95(7)	2.754(6)	3.576(7)	145.5(4)	-x, -y, -z
N2'-H2'O5'	1.05(7)	1.739(6)	2.736(6)	156.3(4)	-1+x, y, z

Analysis of the molecular electrostatic potential distribution [21-23] in 4a-f revealed two distinct minima associated with the =N-N= fragment of the triazine ring and with the carbonyl oxygen in the pyridazine ring, i.e., with atoms capable of acting as proton acceptors in the hydrogen-bond binding of the ligand with the receptor protein. Since the distance between them (approximately 7.7 Å) was quite close to that found analytically in the known triazolopyridazine ligands of the benzodiazepine receptor [24], the compounds were tested in vitro for receptor affinity in competition with ³H-flunitrazepam [25,26]. Inhibition of the labeled ligand binding was, however, very low (<10%). The failure was considered to be caused either by non-planarity of the hydropyridazine ring, which is not a π -electron structure, or by adverse position of the aryl substituent, which may be directed towards an inaccessible region of the receptor protein [27]. Further research is being carried on to account for that problem.

EXPERIMENTAL

General Methods.

Melting points were determined in a Büchi apparatus and are reported uncorrected. The ir spectra were recorded in potassium bromide pellets on a Perkin-Elmer 298 spectrophotometer. ¹H- and ¹³C-nmr spectra were taken with Varian 300 and 200 MHz instruments with TMS as internal standard. Microanalyses were carried out by Mrs. E. Godzisz, Warsaw University of Technology, on a Perkin-Elmer C-H-N analyzer. Merck DC-Plastikfolien with Kieselgel 60 were used in tlc purity checking; the chloroformmethanol-ethyl acetate-saturated aqueous ammonia 3:2:1:0.1 developing system was used. The yield data refer to recrystallized, chromatographically homogeneous compounds.

General Procedure for the Preparation of Ethyl 2-[(4-Aryl-6-oxo-3-tetrahydropyridazinylidene)hydrazono]propanoates 2a-d.

Ethyl pyruvate (0.55 mole) was added to the suspension of the appropriate 4-aryltetrahydropyridazine-3,6-dione 3-hydrazone (0.05 mole) in 100 ml of methanol. Within 10-20 minutes the mixture became homogeneous but another product began to precipitate soon. The mixture was refluxed gently for four hours, cooled and filtered to yield 2. Crude products were purified by recrystallization from ethanol.

Ethyl 2-[(6-Oxo-4-phenyl-3-tetrahydropyridazinylidene)hydrazono]propanoate (2a).

This compound was obtained from 1a as colorless crystals, mp 195-196°, yield 79%; ir: v 3340 and 3180 (NH), 1695 (ester C=O), 1660 (amide C=O) cm⁻¹; ¹H-nmr (deuteriodimethyl sulfoxide): δ 1.44 (t, J = 7.2 Hz, 3H, CH₂CH₃), 2.17 (s, 3H, CH₃), 3.07-3.47 (m, 2H, CH₂), 4.34 (q, J = 7.2 Hz, 2H, CH₂CH₃), 4.81-4.92 (m, 1H, CH), 7.34-7.69 (m, 5H, arom), 10.48 (s, 1H, NH), 10.56 ppm (s,1H, NH).

Anal. Calcd. for C₁₅H₁₈N₄O₃ (302.33): C, 59.59; H, 6.00; N, 18.53. Found: C, 59.62; H, 6.12; N, 18.41.

Ethyl 2-{[4-(4-Chlorophenyl)-6-oxo-3-tetrahydropyridazinylidene]hydrazono}propanoate (2b).

This compound was obtained from 1b as colorless crystals, mp 164-165°, yield 83%;. ir: v 3330 and 3170 (NH), 1690 (ester C=O), 1660 (amide C=O) cm⁻¹; 1 H-nmr (deuteriochloroform): δ = 1.26 (t, J = 7.1 Hz, 3H, CH₂CH₃), 2.03 (s, 3H, CH₃), 3.17-3.38 (m, 2H, CH₂), 4.17 (q, J = 7.1 Hz, 2H, CH₂CH₃), 4.94-5.00 (m, 1H, CH), 7.04-7.33 (m, 4H, arom), 10.53 (s, 1H, NH), 10.59 ppm (s, 1H, NH).

Anal. Calcd. for C₁₅H₁₇ClN₄O₃ (336.78): C, 53.50; H, 5.09; N, 16.64. Found: C, 53.44; H, 5.23; N, 16.41.

Ethyl 2-{[4-(4-Methoxyphenyl)-6-oxo-3-tetrahydropyridazinylidene]hydrazono}propanoate (2c).

This compound was obtained from 1c as colorless crystals, mp 163-164°, yield 78%; ir: v 3330 and 3210 (NH), 1675 (ester C=O), 1640 (amide C=O) cm⁻¹; 1 H-nmr (deuteriochloroform): δ 1.26 (t, J = 7.2 Hz, 3H, CH₂CH₃), 2.04 (s, 3H, CH₃), 2.98-3.29 (m, 2H, CH₂), 3.69 (s, 3H, OCH₃), 4.18 (q, J = 7.2 Hz, 2H, CH₂CH₃), 4.84-4.95 (m, 1H, CH), 6.58-7.38 (m, 4H, arom), 10.59 ppm (broad s, 2H, NH and NH).

Anal. Calcd. for C₁₆H₂₀N₄O₄ (332.36): C, 57.82; H, 6.07; N, 16.86. Found: C, 57.73; H, 6.05; N, 16.80.

Ethyl 2-{[4-(4-Methylphenyl)-6-oxo-3-tetrahydropyridazinylidene]hydrazono}propanoate (2d).

This compound was obtained from 1d as colorless crystals, mp 193-194°, yield 75%; ir: 3330 and 3240 (NH), 1670 (ester C=O), 1640 (amide C=O) cm⁻¹; ¹H-nmr (deuteriochloroform): 1.27 (t, J = 7.2 Hz, 3H, CH₂CH₃), 2.03 (s, 3H, CH₃C=N), 2.21 (s, 3H, CH₃C₆H₄), 3.04-3.31 (m, 2H, CH₂), 4.18 (q, J = 7.2 Hz, 2H, CH₂CH₃), 4.92-5.08 (m, 1H, CH), 6.74-7.43 (m, 4H, arom), 10.67 ppm (broad s, 2H, NH and NH).

Anal. Calcd. for C₁₆H₂₀N₄O₃ (316.36): C, 60.75; H, 6.37; N, 17.71. Found: C, 60.66; H, 6.21; N, 17.68.

General Procedure for Preparation of 2-[(4-Aryl-6-oxo-3-tetrahydropyridazinylidene)hydrazono]propanoic Acids **3a-d**.

The appropriate 2 (0.02 mole) was added with stirring to a sodium ethoxide solution prepared from 0.022 mole of sodium and 60 ml of anhydrous ethanol. The mixture was refluxed 3 hours, cooled, diluted with 100 ml of anhydrous diethyl ether and left standing overnight in a refrigerator. The sodium salt (almost quantitative yield) was filtered off, dissolved in 50 ml of water and the solution acidified under cooling with 20 ml of acetic acid. The filtered product was thoroughly dried (vacuum over phosphorus pentoxide) and purified by recrystallization from nitromethane.

2-[(6-Oxo-4-phenyl-3-tetrahydropyridazinylidene)hydrazono]-propanoic Acid 3a.

This compound was obtained from 2a as colorless crystals, mp 190-191° dec, yield 95%; ir: 3330 and 3210 (NH), 1685 (acid C=O), 1645 (amide C=O) cm⁻¹. 1 H-nmr (deuteriodimethyl sulfoxide): δ 1.93 (s, 3H, CH₃), 2.85-3.19 (m, 2H, CH₂), 4.68-4.82 (m, 1H, CH), 7.19-7.42 (m, 5H, arom), 9.47-10.51 ppm (broad signal, 3H, NH and COOH).

Anal. Calcd. for C₁₃H₁₄N₄O₃ (274.28): C, 56.93; H, 5.14; N, 20.43. Found: C, 56.53; H, 5.22; N, 20.19.

2-{[4-(4-Chlorophenyl)-6-oxo-3-tetrahydropyridazinylidene]-hydrazono}propanoic Acid (3b).

This compound was obtained from 2b as off-white crystals, mp 190-191° dec, yield 85%.

Anal. Calcd. for C₁₃H₁₃ClN₄O₃ (308.72): C, 50.58; H, 4.24; N, 18.15. Found: C, 50.48; H, 4.08; N, 18.05.

2-{[4-(4-Methoxyphenyl)-6-oxo-3-tetrahydropyridazinylidene]-hydrazono}propanoic Acid (3c).

This compound was obtained from 2c as yellowish crystals, mp 194-195° dec, yield 78%.

Anal. Calcd. for $C_{14}H_{16}N_4O_4$ (304.31): C, 55.26; H, 5.30; N,18.41. Found: C, 55.11; H, 5.24; N, 18.48.

2-{[4-(4-Methylphenyl)-6-oxo-3-tetrahydropyridazinylidene]-hydrazono}propanoic Acid (3d).

This compound was obtained from 2d as colorless crystals, mp 196-197° dec, yield 78%.

Anal. Calcd. for C₁₄H₁₆N₄O₃ (288.31): C, 58.32; H, 5.59; N, 19.43. Found: C, 58.36; H, 5.47; N, 19.35.

General Procedure for Preparation of 9-Aryl-8.9-dihydro-3-methyl-4H-pyridazino[6,1-c][1,2,4]triazine-4,7(6H)-diones 4a-d.

A solution of 0.01 mole of 2 in 50 ml of 2-(2-ethoxy-ethoxy)ethanol was refluxed 4 hours and the solvent distilled *in vacuo*. The tan, thick residue was treated with 15 ml of ethanol and cooled in dry ice. The crystalline product was filtered off and recrystallized from ethanol.

An analogous procedure applied to the acid 3a gave a crude mixture consisting of at least 3 compounds (tlc). Upon repeated recrystallization from butanol, 4a was obtained in the yield as low as 20%.

8,9-Dihydro-3-methyl-9-phenyl-4H-pyridazino[6,1-c][1,2,4]-triazine-4,7(6H)-dione (4 \mathbf{a}).

This compound was obtained as colorless crystals, mp 181-182° dec, yield 83% from 2a and 20% from 3a; ir: v 3260 (NH), 1730 and 1680 (C=O) cm⁻¹; ¹H-nmr (deuteriotrifluoroacetic acid): δ 2.67 (s, 3H, CH₃), 3.38-3.77 (m, 2H, CH₂), 4.98-5.27 (m, 1H, CH), 7.30-7.77 ppm (m, 5H, arom).

Anal. Calcd. for C₁₃H₁₂N₄O₂ (256.26): C, 60.93; H, 4.72; N, 21.86; Found: C, 60.82; H, 4.72; N, 21.77.

9-(4-Chlorophenyl)-8,9-dihydro-3-methyl-4H-pyridazino[6,1-c]-[1,2,4]triazine-4,7(6H)-dione (4b).

This compound was obtained from 2b as lightly tan crystals, mp 225-226° dec, yield 75%; ir: v 3160 (NH), 1720 and 1685 (C=O) cm⁻¹; ¹H-nmr (deuteriotrifluoroacetic acid): δ 2.68 (s, 3H, C H_3), 3.32-3.62 (m, 2H, C H_2), 4.94-5.27 (m, 1H, CH), 7.36-7.61 ppm (m, 4H, arom).

Anal. Calcd. for C₁₃H₁₁ClN₄O₂ (290.71): C, 53,71; H, 3.81; N, 19.27. Found: C, 53.67; H, 3.86; N, 19.12.

8,9-Dihydro-9-(4-methoxyphenyl)-3-methyl-4H-pyridazino[6,1-c]-[1,2,4]triazine-4,7(6H)-dione (4c).

This compound was obtained from 2c as lightly tan crystals, mp 199-201° dec, yield 62%; ir: v 3160 (NH), 1705 and 1685 (C=O) cm⁻¹. ¹H-nmr (deuteriotrifluoroacetic acid): δ 2.65 (s, 3H, CH₃), 3.32-3.67 (m, 2H, CH₂), 3.97 (s, 3H, OCH₃), 4.97-5.18 (m, 1H, CH), 6.99-7.54 ppm (m, 4H, arom).

Anal. Calcd. for C₁₄H₁₄N₄O₃ (286.29); C, 58.74; H, 4.93; N, 19.57. Found: C, 58.32; H, 4.71; N, 19.24.

8,9-Dihydro-3-methyl-9-(4-methylphenyl)-4H-pyridazino[6,1-c]-[1,2,4]triazine-4,7(6H)-dione (4d).

This compound was obtained from 2d as colorless crystals, mp 222-224° dec, yield 76%; ir: v 3170 (NH), 1710 and 1685 (C=O) cm⁻¹; ¹H-nmr (deuteriotrifluoroacetic acid): δ 2.41 (s, 3H, CH₃C₆H₄), 2.64 (s, 3H, CH₃C=N), 3.25-3.62 (m, 2H, CH₂), 4.88-5.25 (m, 1H, CH), 7.11-7.38 ppm (m, 4H, arom).

Anal. Caled. for C₁₄H₁₄N₄O₂ (270.29): C, 62.21; H, 5.22; N, 20.73. Found: C, 62.06; H, 5.13; N, 20.54.

6-Benzyl-8,9-dihydro-3-methyl-9-phenyl-4H-pyridazino[6,1-c]-[1,2,4]triazine-4,7(6H)-dione (5).

Tetrabutylammonium bromide (0.13 g, 0.0004 mole), powdered potassium hydroxide (0.1 g, 0.002 mole) and benzyl bromide (0.39 g, 0.0023 mole) were added to the solution of 0.5 g (0.002 mole) of 4a in 7.5 ml of anhydrous DMF. The mixture was stirred 6 hours at room temperature, the solvent was distilled *in vacuo* and the oily residue was treated with chloroform. Filtration removed some inorganic salts and the filtrate was repeatedly washed with saline and water. Evaporation of the chloroform left a yellowish oil which slowly solidified. Recrystallization from ethanol yielded 46% of 5 as colorless crystals, mp 148-150; ir: v 1700 and 1685 (C=O) cm⁻¹; ¹H-nmr (deuteriotrifluoroacetic acid): δ 2.65 (s, 3H, CH₃), 3.40-3.75 (m, 2H, endocyclic CH₂), 4.62-4.84 (m, 1H, CH), 5.08 (d, 2 J =15.7 Hz, 1H, benzyl CH₂), 5.29 (d, 2 J =15.7 Hz, 1H, benzyl CH₂), 7.38-7.69 ppm (m, 10H, arom).

Anal. Calcd. for C₂₀H₁₈N₄O₂(346.39): C, 69.35; H, 5.24; N, 16.17. Found: C, 69.22; H, 5.29; N, 16.21.

General Procedure for Preparation of Ethyl α -[(6-Oxo-4-phenyl-3-tetrahydropyridazinylidene)hydrazono]arylacetates **2e-f**.

Ethyl aroylformate (0.011 mole) and 1a (0.023 mole) were dissolved in 2 ml of methanol and left standing at room temperature for 7 days and next at 0° for 5 days. The precipitated product was recrystallized from methanol.

Ethyl α -[(6-Oxo-4-phenyl-3-tetrahydropyridazinylidene)hydrazono]phenylacetate (2e).

This compound was obtained from 1a and ethyl benzoylformate as colorless crystals, mp 192-194°, yield 30%; ir: v 3290, 3180 and 3120 (NH), 1700 and 1670 (C=O) cm⁻¹; ¹H-nmr (deuteriochloroform): δ 1.31 (t, J = 7.2 Hz, 3H, CH₂CH₃), 2.57-2.80 (m, 2H, CHCH₂), 4.27 (q, J = 7.2 Hz, 2H, CH₂CH₃), 4.67-4.80 (m, 1H, CHCH₂), 7.04-7.50 ppm (m, 10H, arom).

Anal. Calcd. for $\overline{C}_{20}H_{20}N_4O_3$ (364.40): C, 65.92; H, 5.53; N, 15.37. Found: C, 65.70; H, 5.57; N,15.30.

Ethyl α -[(6-Oxo-4-phenyl-3-tetrahydropyridazinylidene)hydrazono]-3-nitrophenylacetate (2f).

This compound was obtained from 1a and ethyl 3-nitrobenzoylformate as light yellow crystals, mp 167-169°, yield 35%; ir: v 3290 and 3190 (NH), 1720 and 1670 (C=O) cm⁻¹; ¹H-nmr (deuteriochloroform): δ 1.30 (t, J = 7.2 Hz, 3H, CH₂CH₃), 2.42-2.79 (m, 2H, CHCH₂), 4.24 (q, J = 7.2 Hz, 2H, CH₂CH₃), 4.62-4.89 (m, 1H, CHCH₂), 7.29-8.11 ppm (m, 9H, arom).

Anal. Calcd. for C₂₀H₁₉N₅O₅ (409.40): C, 58.68; H, 4.68; N, 17.11. Found: C, 58.36; H, 4.70; N, 16.97.

General Procedure for Preparation of 3-Aryl-8,9-dihydro-9-phenyl-4*H*-pyridazino[6,1-*c*][1,2,4]triazine-4,7(6*H*)-diones 4e-f.

These compounds were prepared from 2e-f analogously as described for 4a-d or directly from 1a and the appropriate aroylformate as follows: 1a (0.0023 mole) and 0.0112 mole of ethyl aroylformate were refluxed for 30 minutes in 5 ml of 2-(ethoxyethoxy)ethanol. The crystals which separated on cooling were filtered off and purified by recrystallization.

8,9-Dihydro-3,9-diphenyl-4H-pyridazino[6,1-c][1,2,4]triazine-4,7(6H)-dione (4e).

This compound was obtained as lightly yellow crystals, mp 251-251° (nitromethane), yield 85% from 2e and 56% from 1a; ir: v 3180 (NH), 1715 and 1680 (C=O) cm⁻¹; ¹H-nmr (deuteriotrifluoroacetic acid): δ 3.07-3.54 (m, 2H, CH₂), 4.85-5.30 (m, 1H, CH), 7.12-8.15 ppm (m, 10H, arom).

Anal. Calcd. for C₁₈H₁₄N₄O₂ (318.33): C, 67.92; H, 4.43; N, 17.60. Found: C, 67.80; H, 4.42; N, 17.51.

8,9-Dihydro-3-(3-nitrophenyl)-9-phenyl-4H-pyridazino[6,1-c]-[1,2,4]triazine-4,7(6H)-dione (4 \mathbf{f}).

This compound was obtained as yellow crystals, mp 250-251° (methanol), yield 55% from 2f and 38% from 1a; ir: v 3190 (NH) 1710 and 1685 (C=O) cm⁻¹; 1 H-nmr (deuteriotrifluoroacetic acid): δ 3.20-3.52 (m, 2H, CH₂), 4.85-5.19 (m, 1H, CH), 7.25-8.02 ppm (m, 9H, arom).

Anal. Calcd. for C₁₈H₁₃N₅O₄ (363.33): C, 59.50; H, 3.61; N, 19.28. Found: C, 59.65; H, 3.79; N, 19.30.

General Procedure for Preparation of Ethyl 3-[(4-Aryl-6-oxo-3-tetrahydropyridazinylidene)hydrazono]butanoates **6a-d**.

A mixture of 0.05 mole of the appropriate 1 and 0.075 mole (9.76 g) of ethyl acetoacetate in 100 ml of methanol was refluxed gently for 4 hours. Evaporation of the solvent left a solid product which was washed with a small volume of thoroughly cooled ethyl ether and then with hexane and finally recrystallized to yield 6.

Ethyl 3-[(6-Oxo-4-phenyl-3-tetrahydropyridazinylidene)hydrazono]butanoate (6a).

This compound was obtained from 1a as colorless crystals, mp 139-141° (nitromethane), yield 92%; ir: v 3240 and 3255 (NH), 1740 (ester C=O), 1660 (amide C=O) cm⁻¹; ¹H-nmr

(deuteriochloroform) of the major isomer: δ 1.20 (t, $^{3}J = 7.15$ Hz, 3H, CH_2CH_3), 1.88 (s, 3H, = CCH_3), 2.79 and 2.81 (2 dd, 2 J = 16.7 Hz, 3 J = 3.8 and 7.5 Hz, 2 non-equivalent H, endocyclic CH₂), 3.20 (s, 2H, =CCH₂), 4.08 and 4.10 (4 dd, $^{2}J =$ 11.0 Hz, ${}^{3}J = 7.2$ Hz, 2 non-equivalent H, $CH_{2}CH_{3}$), 4.73 (m, 1H. CH), 7.29-7.32 (m, 5H, arom), 9.43 (s, 1H, NH), 10.23 ppm (s, 1H, NH); ¹³C-nmr (deuteriochloroform) of the major isomer: δ 14.1 (CH₂CH₃), 15.9 (=CCH₃), 34.2 (endocyclic CH₂), 36.7 (CH), 44.7 (=CCH₂), 60.9 (CH₂CH₃), 127.4, 127.6, and 129.0 (arom CH), 137.5 (arom C), 143.9 (exocyclic C=N), 154.7 (endocyclic C=N), 167.3 (ester C=O), 170.0 ppm (pyridazine C=O); ¹H-nmr (deuteriochloroform) of the minor isomer: δ 1.17 $(t, {}^{3}J = 7.2 \text{ Hz}, 3H, CH_{2}CH_{3}), 1.98 \text{ (s, 3H, =CC}H_{3}), 2.81 \text{ and}$ 2.83 (2 dd, 2 non-equivalent H, $^{2}J = 16.8$ Hz, $^{3}J = 3.9$ and 7.4 Hz, endocyclic CH₂), 3.25 (s, 2H, =CCH₂), 4.04 (q, ${}^{3}J$ = 7.2 Hz, 2H, CH₂CH₃), 4.55 (m, 1H, CH), 7.29-7.32 ppm (m, 5H, arom); ¹³C-nmr (deuteriochloroform) of the minor isomer: 14.1 (CH₂CH₃), 24.3 (=CCH₃), 34.4 (endocyclic CH₂), 37.5 (CH), 46.3 (=CCH₂), 61.5 (CH₂CH₃), 127.5, 127.7, and 129.1 (arom CH), 137.36 (arom C), 143.9 (exocyclic C=N), 153.7 (endocyclic C=N), 167.1 (ester C=O), 168.3 ppm (pyridazine C=O).

Anal. Calcd. for C₁₆H₂₀N₄O₃ (316.36): C, 60.75; H, 6.37; N, 17.71. Found: C, 60.67; H, 6.31; N, 17.88.

Ethyl 3-{[4-(4-Chlorophenyl)-6-oxo-3-tetrahydropyridazinylidene]hydrazono}butanoate (6b).

This compound was obtained from **2b** as colorless crystals, mp 146-148° (ethanol), yield 75%; ir: v 3260 (NH), 1740 (ester C=O), 1665 (amide C=O) cm⁻¹; ¹H-nmr (deuteriochloroform): δ 1.15 (t, 3H, CH₂CH₃), 1.88 (s, 3H, =CCH₃), 2.63-2.78 (m, 2H, endocyclic CH₂), 3.08 (s, 2H, =CCH₂), 3.97 (q, 2H, CH₂CH₃), 4.71-4.76 (m, 1H, CH), 7.12-7.37 (m, 4H, arom), 9.93 (s, 1H, NH), 10.63 ppm (s, 1H, NH).

Anal. Calcd. for C₁₆H₁₉ClN₄O₃ (350.80): C, 54.78; H, 5.46; N, 15.97. Found: C, 55.02; H, 5.50; N, 15.89.

Ethyl 3-{[4-(4-Methoxyphenyl)-6-oxo-3-tetrahydropyridazinylidene]hydrazono}butanoate (6c).

This compound was obtained from 2c as colorless crystals, mp 184-185° (ethanol-2-methoxyethanol 1:3), yield 91%; ir: v 3240 (NH), 1735 (ester C=O), 1655 (amide C=O) cm⁻¹; ¹H-nmr (deuteriochloroform): δ 1.18 (t, 3H, CH₂CH₃), 1.90 (s, 3H, =CCH₃), 2.61-2.80 (m, 2H, endocyclic CH₂), 3.15 (s, 2H, =CCH₂), 3.60 (s, 3H, OCH₃), 3.84 (q, 2H, CH₂CH₃), 4.70-4.76 (m, 1H, CH), 6.92-7.04 (m, 4H, arom), 9.83 (s, 1H, NH), 10.52 ppm (s, 1H, NH).

Anal. Calcd. for C₁₇H₂₂N₄O₄ (346.39): C, 58.95; H, 6.40; N, 16.17. Found: C, 58.79; H, 6.57; N, 15.97.

Ethyl 3-{[4-(4-Methylphenyl)-6-oxo-3-tetrahydropyridazinylidene]hydrazono}butanoate (6d).

This compound was obtained from 2d as colorless crystals, mp 149-150° (ethanol), yield 71%; ir: \vee 3250 (NH), 1735 (ester C=O), 1670 (amide C=O) cm⁻¹; ¹H-nmr (deuteriochloroform): δ 1.13 (t, 3H, CH₂CH₃), 1.84 (s, 3H, =CCH₃), 2.22 (s, 3H, C₆H₄CH₃), 2.63-2.76 (m, 2H, endocyclic CH₂), 3.11 (s, 2H, =CCH₂), 4.00 (q, 2H, CH₂CH₃), 4.62-4.74 (m, 1H, CH), 6.97-7.09 (m, 4H, arom), 9.73 (s, 1H, NH), 10.49 ppm (s, 1H, NH).

Anal. Calcd. for C₁₇H₂₂N₄O₃ (330.39): C, 61.80; H, 6.71; N, 16.96. Found: C, 61.66; H, 6.60; N, 16.89.

General Procedure for Preparation of 3-(2,4-dihydro-5-methyl-3-oxo-3*H*-pyrazol-2-yl)-4-aryl-4,5-dihydropyridazin-6(1*H*)-ones (7a-d).

The appropriate 6 (0.02 mole) was added portionwise to a sodium ethoxide solution prepared from 0.51 g (0.022 mole) of sodium and 50 ml of ethanol. The mixture was refluxed with stirring for 3 hours, cooled and filtered. The obtained sodium salt (approximately 90% yields) was dissolved in 50 ml of water and the solution acidified with 2N hydrochloric acid. The crude product was dissolved in methanol and the solution was passed through a silica gel column. Recrystallization from methanol gave pure products.

3-(2,4-dihydro-5-methyl-3-oxo-3*H*-pyrazol-2-yl)-4,5-dihydro-4-phenylpyridazin-6(1*H*)-one (7a).

This compound was obtained from **6a** as colorless crystals, mp 209-210°, yield 92%; ir: v 3170 (NH), 1665 and 1630 (C=O) cm⁻¹; 1 H-nmr (deuteriodimethyl sulfoxide): δ 2.08 (s, 3H,CH₃), 2.55 (s, 2H, CH₂), 3.07-3.13 (m, 2H, CHCH₂), 5.41-5.47 (m, 1H, CHCH₂), 7.19-7.36 (m, 5H, arom), 10.72 ppm (s,1H, NH).

Anal. Calcd. for $C_{14}H_{14}N_4O_2$ (270.29): C, 62.21; H, 5.22; N, 20.73; Found: C, 62.07; H, 5.20; N, 20.68.

3-(2,4-Dihydro-5-methyl-3-oxo-3*H*-pyrazol-2-yl)-4-(4-chlorophenyl)-4,5-dihydropyridazin-6(1*H*)-one (7b).

This compound was obtained from **6b** as colorless crystals, mp 210-211°, yield 82%; ir: v 3170 (NH), 1670 and 1620 (C=O) cm⁻¹; ¹H-nmr (deuteriodimethyl sulfoxide): δ 2.08 (s, 3H, CH₃), 2.55 (s, 2H, CH₂), 3.07-3.15 (m, 2H, CHCH₂), 5.42-5.47 (m, 1H, CHCH₂), 7.20-7.43 (m, 4H, arom), 10.75 ppm (s, 1H, NH).

Anal. Calcd. for C₁₄H₁₃ClN₄O₂ (304.74): C, 55.18; H, 4.30; N, 18.39. Found: C, 55.23; H, 4.37; N, 18.49.

3-(2,4-Dihydro-5-methyl-3-oxo-3*H*-pyrazol-2-yl)-4,5-dihydro-4-(4-methoxyphenyl)pyridazin-6(1*H*)-one (7c).

This compound was obtained from 6c as colorless crystals, mp 229-230°, yield 95%; ir: v 3160 (NH), 1665 and 1615 (C=O) cm⁻¹; 1 H-nmr (deuteriodimethyl sulfoxide): δ 2.08 (s, 3H, CH₃), 2.56 (s, 2H, CH₂), 3.04-3.08 (m, 2H, CHCH₂), 3.72 (s, 3H, OCH₃), 5.35-5.41 (m, 1H, CHCH₂), 6.88-7.17 (m, 4H, arom), 10.70 ppm (s, 1H, NH).

Anal. Caled. for C₁₅H₁₆N₄O₃ (300.32): C, 59.99; H, 5.37; N, 18.66. Found: C, 59.88; H, 5.37; N, 18.57.

3-(2,4-Dihydro-5-methyl-3-oxo-3*H*-pyrazol-2-yl)-4,5-dihydro-4-(4-methylphenyl)pyridazin-6(1*H*)-one (7**d**).

This compound was obtained from 6d as colorless crystals, mp 199-200°, yield 91%; ir: v 3170 (NH), 1665 and 1620 (C=O) cm⁻¹; 1 H-nmr (deuteriodimethyl sulfoxide): δ 2.07 (s, 3H, pyrazole CH₃), 2.25 (s, 3H, C₆H₄CH₃), 2.56 (s, 2H, CH₂), 3.04-3.10 (m, 2H, CHCH₂), 5.35-5.41 (m, 1H, CHCH₂), 7.06-7.15 (m, 4H, arom), 10.69 (s, 1H, NH).

Anal. Caled. for C₁₅H₁₆N₄O₂ (284.32): C, 63.37; H, 5.67; N, 19.71. Found: C, 63.24; H, 5.66; N, 19.81.

Crystal X-ray Analysis of 7d.

Crystals were grown from ethanol. Crystallographic measurements were performed on an Enraf-Nonius CAD-4 diffractometer using $\text{Cu}K_{\alpha}$ radiation ($\lambda=1.54178~\text{Å}$). The $\omega/2\Theta$ scan technique was applied with 2Θ 150°. Three reflections monitored

every 100 reflections were used as standards and remeasured during the data collection; there was no indication of crystal decomposition. A single crystal of approximate dimensions 0.2 x 0.2 x 0.3 mm was used for intensity data collection with a total of 1101 reflections measured (-7 \leq h \leq 6, -13 \leq k \leq 12, -15 \leq 1 \leq 0). A total of 1054 (R_{int} = 0.058) independent data were classified as observed (F > 4 α (F)) and Lorentz polarization corrections were applied to these data. The cell dimensions were obtained and refined by the least-square method involving 25 reflections (10 \leq Θ \leq 42). The structure was solved by direct methods using the SHELXS-86 program [28]. It was refined on F² by the full-matrix-block least-squares method [29]. The positions of all H atoms were obtained from difference maps calculated at the completion of the anisotropic refinement.

Crystal Data of 7d.

Mol. formula $C_{15}H_{16}N_4O_2$, mol. mass 284.32, triclinic, PĪ space group; unit cell dimensions: a=5.7690(10) Å, b=10.160(2) Å, c=12.335(2) Å; = 79.56(3), = 85.93(3), = 84.14(3); V=706.3(2) ų; $d_{\rm X}=1.337$ Mgm⁻³; Z=2; F(000)=300; $\mu({\rm Cu}K_{\alpha})=0.753$ mm⁻¹. Final R=0.058 for 1054 independent reflections [20].

Determination of the Affinity for the Benzodiazepine Receptor.

Rat brains were homogenized at 0° in 20 volumes of 50 mM tris hydrochloride (pH 7.4) and the homogenates were incubated 1 hour at 37° and centrifuged at 20,000 G. Samples of the homogenate (800 μ l corresponding to 13.3 mg of brain tissue) were mixed with 100 μ l of ³H-flunitrazepam (specific activity 81 Ci/mmole) and 100 μ l of the tested compound (10^{-6} M). The mixture was incubated 2 hours at 5° and filtered through a glass fiber filter (Whatman GF/C). The filter was washed twice with 5- μ l portions of tris, dried, and immersed in 10 ml of the scintillation liquid [1,4-bis(5-phenyl-2-oxazolyl)benzene 50 mg, 2,5-diphenyloxazole 4 g, methanol 20 ml, and toluene 1000 ml]. Radioactivity was measured in a Betamatic II Kontron β -scintillation counter and expressed as % inhibition of binding of the labeled ligand.

Acknowledgement.

A major part of this research was carried out under grant # 033/P05/95/09 from the Committee of Scientific Research, Poland. This financial support is gratefully acknowledged. We wish also to thank Dr.T.Jagodzinski, Technical University, Szczecin, Poland, for taking the HETCOR nmr spectra.

REFERENCES AND NOTES

[1] Part 3. J. Karolak-Wojciechowska, H. B. Trzezwinska, J. Lange, and M. Wieczorek, *Acta Cryst.*, C51, 1829 (1995).

[2] R. F. Squires, D. I. Benson, C. Braestrup, J. Coupet, C. A. Klepner, V. Myers, and B. Beer, *Pharmacol. Biochem. Behav.*, 10, 825 (1979).

[3] A. S. Lippa, D. J. Critchett, M. V. Sano, C. A. Klepner, E. N. Greenblatt, J. Coupet, and B. Beer, *Pharmacol. Biochem. Behav.*, **10**, 831 (1979).

[4] J. D. Albright, D. B. Moran, W. B. Wright, Jr., J. B. Collins, B. Beer, A. S. Lippa, and E. N. Greenblatt, J. Med. Chem., 24, 592 (1981).

[5] C. R. Gardner, Drug Dev. Res., 12, 1 (1989).

[6] C. A. Klepner, A. S. Lippa, D. I. Benson, M. C. Sano, and B. Beer, *Pharmacol. Biochem. Behav.*, 11, 457 (1979).

- [7] W. Sieghart and M. Karobath, Nature, 286, 285 (1980).
- [8] W. S. Young III, D. Niehoff, M. J. Kuhar, B. Beer, and A. S. Lippa, *J. Pharmac. Exp. Ther.*, **216**, 425 (1981).
- [9] A. Doble and I. L. Martin, Trends Pharmacol. Sci., 13, 76 (1992).
- [10] W. Zhang, K. F. Koehler, P. Zhang, and J. M. Cook, *Drug Design Discovery*, 12, 193 (1995).
- [11] J. Lange and H. Tondys, Polish Patent 125,777 (1979); Chem. Abstr., 104, 88609e (1986).
- [12] J. Lange, H. Tondys, W. Koberda, and M. Gniewosz, Synth. Commun., 23, 1371 (1993).
- [13] The compounds 2, 3 and 6 are named throughout the paper as the azine tautomers derived from 4-aryltetrahydropyridazine-3,6-dione 3-hydrazones (1) and the appropriate keto esters.
- [14] J. Lange, J. Karolak-Wojciechowska, M. Gniewosz, and J. Plenkiewicz. *Pharmazie*, 49, 21 (1994).
- [15] B. V. Alaka, D. Patnaik, and M. K. Rout, J. Indian Chem. Soc., 59, 1168 (1982).
- [16] K. C. Joshi and P. Chand, Heterocycles, 16, 43 (1981).
- [17] M. Dorneanu, E. Stefanescu, and G. Grosu, Rev. Med.-Chir., 91, 541 (1987); Chem. Abstr., 109, 170343a (1988).

- [18] H. Zimmer and A. Amer, Heterocycles, 26, 1177 (1987).
- [19] S. N. Sawhney, P. K. Sharmla, and A. Gupta, *Indian J. Chem.*, 31B, 421 (1992).
- [20] Further details of the crystal structure of 7d, including structure factors, anisotropic displacement parameters, and H-atom coordinates, are available on request from one of the authors (J. K.-W.).
- [21] MOPAC.6 Program Packet, Quantum Chemistry Program Exchange, No. 326.
- [22] J. J. P. Steward, J. Comp. Chem., 10, 209 (1989).
- [23] W. Kwiatkowski and J. Karolak-Wojciechowska, SAR QSAR Envir. Res., 1, 233 (1993).
- [24] J. Karolak-Wojciechowska, J. Lange, W. Kwiatkowski, M. Gniewosz, and J. Plenkiewicz, *Bioorg. Med. Chem.*, 2, 773 (1994).
- [25] R. Squires and C. Braestrup, Nature, 266, 732 (1977).
- [26] H. Möhler and T. Okada, Science, 198, 849 (1977).
- [27] H. Diaz-Arauzo, K. F. Koehler, T. J. Hagen, and J. M. Cook, Life Sci., 49, 207 (1991).
- [28] G. M. Sheldrick, SHELXS-86 The Programme for Solution of Crystal Structures, University of Göttingen, Germany, 1986.
- [29] G. M. Sheldrick, SHELXTL/PC User's Manual, Siemens Analytical X-ray Instruments, Inc., Madison, WI, USA (1990).